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(71)Applicant: TOHOKU PIONEER CORP

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(72)Inventor: SHIRAHATA KUNIHIKO

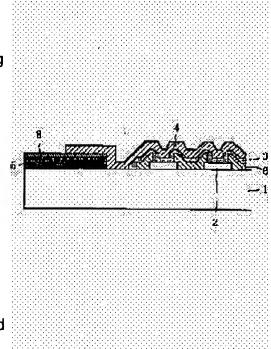
NAGAYAMA KENICHI

(54) ORGANIC ELECTROLUMINESCENT DISPLAY DEVICE AND ITS MANUFACTURING METHOD

(57)Abstract:

PROBLEM TO BE SOLVED: To provide an organic electroluminescent display device, capable of preventing generation of an alteration layer 7 causing contact resistance to increase by forming a barrier layer 8 as a surface layer of an extraction electrode 5, and being driven by a comparatively constant voltage.

SOLUTION: In this organic electroluminescence display device, the barrier layer 8 is formed as the surface layer of the extraction electrode 5 installed on a transparent substrate 1. The extraction electrode 5 is brought into contact via the barrier layer 8 with a metal electrode 4 laminated on the transparent substrate 1 via an organic luminescent layer 3. The extraction electrode 5 is formed from metallic conductive material such as Cr, Al, Cu, Ag,



Au, Pt, Pd, Ni, Mo, Ta, Ti, W, C, Fe, In, Ag-Mg, Zn or the like. The barrier layer 8 is formed of a metal, having satisfactory heat-resistant alteration property and a high-melting point, a noble metal, an oxide, a nitride, or an oxidized nitride.

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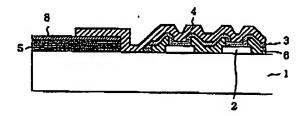
(21)出窟番号	特数2000—171946(P2000—171946)	(71) 出題人 000221926
		東北ノイオニア株式会社
(22)出版日	平成12年6月8日(2000.6.8)	山形県天政市大字久野本字日光1105級地
		(72) 発明者 白幡 邦彦
		山形界米沢市八幡原四丁目3146番7号 東
		北バイオニア株式会社内
	•	(72) 発明者 永山 健一
•		山形県米沢市八幡原四丁目3148番7号 東
•		北バイオニア株式会社内
		(74)代理人 100092392
		弁理士 小倉 耳
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(54) 【発明の名称】 有機エレクトロルミネッセンス表示デバイス及びその製造方法

(57) [要約]

【目的】 引出し電揺5の表層にパリア層8を形成する ことにより、接触抵抗を上昇させる変質層7の生成を防止し、比較的定電圧で駆動可能な有機エレクトロルミネッセンス表示デバイスを得る。

【構成】 この有機エレクトロルミネッセンス表示デバイスは、透明基板1に設けられた引出し電極5の表層にパリア層8を形成している。引出し電極5は、有機発光層3を介して透明基板1上に積層された金属電極4にパリア層8を介して接触している。引出し電極5は、Cr, Al, Cu, Ag, Au, Pt, Pd, NJ, Mo, Ta, Ti, W, C, Fe, In, Ag-Mg, 2n等の金属質導電材料で形成される。パリア層8は、耐熱変質性の良好な高融点金属、費金属、酸化物、窒化物又は酸窒化物で形成される。



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【特許請求の範囲】

【請求項1】 透明基板に設けられた引出し電極の表層 にパリア層を形成し、有機発光層を介して透明基板上に 積層された金属性極に、前記パリア層を介して前記引出 し電極が接触していることを特徴とする有機エレクトロ ルミネッセンス表示デバイス。

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【請求項2】 Cr, Al, Cu, Ag, Au, Pt, Pd, Ni, Mo, Ta, Ti, W, C, Fe, In, Ag-Mg, Znから選ばれた金属質導電材料で引出し電極が形成されている請求項1記載の有機エレクトロル 10ミネッセンス表示デバイス。

【蔚求項3】 耐熱変質性の良好な高融点金属、貴金 属、酸化物、窒化物又は酸窒化物でパリア層が形成され ている防求項1記載の有機エレクトロルミネッセンス表 示デバイス。

【請求項4】 引出し電極用金属質導電材料の酸化物、 窒化物又は酸窒化物の薄層としてパリア層が形成されて いる請求項1記載の有機エレクトロルミネッセンス表示 デバイス。

【請求項5】 引出し電極用金属質導電材料の酸化物、 空化物又は酸空化物の溶層を密着性改善層として介在さ せて引出し電極が適明基板上に形成されている請求項1 記載の有機エレクトロルミネッセンス表示デバイス。

【請求項6】 バリア層及び密着性改善層を含む引出し 電極を同じエッチング液で加工することを特徴とする有 徴エレクトロルミネッセンス表示デバイスの製造方法。

【発明の詳細な説明】

[0001]

【産業上の利用分野】本発明は、電務に対する接触抵抗 を低く総持し、比較的低い低圧で駆動可能な有機エレク 30 トロルミネッセンス表示デバイスに関する。

[0002]

【従来の技術】ブラウン管に代わるカラー表示装置として、被品ディスプレイパネルが多方面で替及している。しかし、被品ディスプレイパネルは、パックライトから被品層を通過した光で函像表示する方式であるため、見る角度や周囲の明暗度によって函像が見難くなることがある。この点、面発光によって必要画像を表示するエレクトロルミネッセンス索子を用いたディスプレイパネルは、四像の見易さが観察角度によって変わることなく、暗所でも十分な鮮明度で画像が観察される。

(0003) エレクトロルミネッセンス材料として種々の無機材料及び有機材料が知られているが、低電力で高輝度発光する有機材料が注目されている。有機発光材料を使用した表示デバイスは、透明基板1に複数のストライプ状逸明電極2、直接発光層3、透明電極2に直交する複数のストライプ状金属電極4を崩次積層した構造(図1)をもっている。透明基板1の上には、金属電極4に導過する引出し電極5が積層されている。

[0004] 適明電磁2及び金属電極4で形成されるX 50

Yマトリックス上の所定位置に駆動電流を供給すると、 階極側からのホールと陰極関からの電子が有機発光層 3 で再結合し、有機発光体分子が励起され面状に発光する。発光は、適明電極 2 及び透明器板 1 を通して外部に 取り出される。本出頭人は、この有機エレクトロルミネッセンス表示デバイスにおいて層間絶縁原 6 を形成する とき、週明電極 2 エッジ部の段差に起因する適明電極 2 と金属電極 4 との短絡が防止され、鮮明な画像が表示されることを特開 平8 - 3 1 5 9 8 1 号公報で紹介した。

[0005]

【発明が解決しようとする課題】層問絶縁職6は、たとえばシリカ分散ペーストを所定パターンで蟄布し、ペークによって形成される。或いは、ポリイミド系強料を強布して、熱処理で契約させることにより安定なポリマー皮膜として形成される。何れの場合も、層間絶縁度6の形成にペークを必要とし、引出し電極5の表面を変質させやすい。引出し電極5の材料としては抵抗値が低い金属が使用されているが、ペークによって表面に変質層7(図2)が生成すると、金属電極4との接触抵抗Rが増加する。その結果、有機エレクトロルミネッセンス表示デバイスの駆動に高い電圧が必要となり、また昇遠による有機エレクトロルミネッセンス表示デバイスの場合に追される。

[0006]

【課題を解決するための手段】本発明は、このような問題を解消すべく紊出されたものであり、引出し電極の表面にパリア層を形成することにより、ベーク等の加熱時に引出し電極の変質、ひいては引出し電極と金属電極との接触抵抗の上昇を防止し、比較的低い電圧で駆助可能な有機エレクトロルミネッセンス表示デバイスを提供することを目的とする。本発明の有機エレクトロルミネッセンス表示デバイスは、その目的を選成するため、透明基板に設けられた引出し段極の表層にパリア層を形成し、有機発光層を介して透明基板上に積層された金属電極に、前記パリア層を介して前記引出し電極が接触していることを特徴とする。

【0007】引出し電極は、たとえばCr, Al, Cu, Ag, Au, Pt, Pd, Ni, Mo, Ta, Tl, W, C. Fe, In, Ag-Mg, 2nから選ばれた金属質導電材料で形成される。パリア層は、耐熱変質性の良好な高融点金属、貴金属、酸化物、窒化物又は酸窒化物で形成される。また、引出し塩極用金属質等電材料の酸化物、窒化物又は酸窒化物の薄層としてもパリア層を形成できる。更に、引出し電極用金属質導電材料の酸化物、窒化物又は酸窒化物の薄層を密着性改替層として形成した後で、引出し電極を選明基板上に形成してもよい。引出し電極の表層に形成されるパリア層は、密着性改善層を含む引出し電極と共に同じエッチング液を用いて加工することができる。

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【実施の形成】本発明に従った有機エレクトロルミネッ センス表示デバイスは、たとえば図3に示すように、引 出し電極5の裏面をパリア層8で被覆している。その他 は、従来の有機エレクトロルミネッセンス表示デバイス (図1) とほぼ同じ層構造をもっている。パリア層8の 形成によって、層間絶縁膜6形成時に引出し電極5の表 面が熱的に保護され、変質層7の生成が防止される。引 出し電極5の材料には、たとえば抵抗値の低いCェ、A 1. Cu. Ag. Au, Pt. Pd. Ni, Mo, T a, Tj, W, C, Fe, Jn, Ag-Mg, 2n等の i0 金属又は合金が使用される。 これらの金属又は合金は、 戎若, スパッタリング, イオンプレーティング等の方法 により菊膜として透明基板1上に形成される。

[0009] バリア暦8の材料としては、層間絶縁膜6 形成時のベーク等、熱処理の際に変質しない高融点金 属、黄金属、酸化物、衰化物、酸窒化物等が適してい る。 導電性を妨げない程度の膜厚(具体的には10~1 00人、好ましくは10~500人、最適には10~2 00点) でパリア層8を形成する場合、絶録物に分類さ れている材料でも使用可能である。具体的には、耐変質 20 性に優れたAu. Pt, Pd, W, Mo等の企民材料の 他に、CrO., AlO., MoO., FeO., Ni On Agon Znon Taon Won Sion S no, CrN, SiN, CrN,O, ŤiC, Ta SnO, TaSnO,N,等の酸化物、窒化物、酸窒化 物等がある。

【0010】また、引出し電極5に使用する金属又は合 金の酸化物、空化物又は酸窒化物をパリア層8に使用す ることも可能である。この場合、たとえば引出し電極 5 を成職する最終段階で、雰囲気に酸素及び/又は窒素を 30 導入することにより金属酸化物、金属窒化物、金属酸密 化物等のパリア暦8が引出し電極5上に形成される。引 出し電極5と同じ金属の酸化物、変化物又は酸窒化物で 形成されたパリア層8は、引出し電極5と同じエッチン グ波を用いてパターニングできる場合も多い。引出し電 極5の表層に形成されたパリア層8は、層間絶縁膜6を 形成するペーク処理時に変質層7の生成を防止する上で 有効であるが、ベーク処理時に限らず他の熟処理工程に おいても変質層?の生成を有効に抑制し、金属電極4に 対する引出し電極5の接触抵抗を低位に維持する。

【0011】 質に、引出し電極5の形成に先立って、酸 森を導入したAr 雰囲気でたとえばCr をスパッタリン グすると、CrO.質の薄膜が透明基板1上に形成され る。CrO,質抑膜は、基板1に対する引出し電極5の 密着性を向上させる密着性改善層9(図5)として側 く。CrO.質物膜も、引出し電極5と同じエッチング 液を用いてパターニングできる。

[0012]

【実施例1】透明基板1に透明電極2を形成した後、A r 雰囲気でCrをターゲットに用いたスパッタリングに 50 基板1の上に形成した後、金属電極4を堆積させた。形

よって順厚2900人の純Cr海膜を形成した(図4 a)。次いで、Ar雰囲気に酸素を導入し、CrOiを 引出し危極5の上に堆積させ、膜厚100人のCrO。 郊膜を形成した (図4b)。 CrO, 蒋膜が形成された 後の透明基板1にレジストパターンを形成した後、硝酸 セリウムアンモンを含むエッチング液を用いてエッチン グし、レジストを刺激することによって純Cェ苺版及び CrO、薄膜をパターニングして所定形状の引出し電極 5 及びパリア暦 8 を形成した(図4 c)。このとき、同 じエッチング液で引出し低極5及びパリア層8の双方を パターニングできるため、簡便な工程で所定形状の引出 し電極5及びパリア暦8が作成できる。

【0013】次いで、逸明恭板1をオープンに装入して 150~300℃の概囲でベークした後、A1を蒸着す るりフトオフ法で金属電極4を形成した(図4d)。形 成された金属電極4と引出し電極5との接触抵抗Rを測 定した結果を設1に示す。表1から明らかなように、C. rO。をパリア層8として形成した引出し電極5では、 ペーク後にも金属電極4に対して低い接触抵抗を維持し ており、ペーク温度200℃以上でほぼ一定した低接触 抵抗を示した。

[0014]

【実施例2】透明基板1に対する引出し電極5の密着性 を向上させるため、酸素を導入したAr雰囲気でCrを ターゲットにしたスパッタリングによって旗厚100Å のCr〇,蒋膜(密着性改善層9)を透明基板1上に形 成した(図5a)。次いで、資ネフリーのAr毎囲気に 代えてCrをスパッタリングし、鎮原2800人のCr 蒋晓(引出し電極5)を堆積させ(図5b)、再び酸素 をAr 雰囲気に導入してスパッタリングし、膜厚100 AのCrOx荷膜(パリア暦8)を金属電極4の上に形 成した(図5c)。各符膜を実施例1と同じエッチング 方法でパターニングし、下層に密着性改善用9. 表層に バリア暦8を備えた引出し電極5を形成した(図5) d)。次いで、透明基板1をオープンに装入して150 ~300℃の範囲でペークした後、A1を蒸着するリフ トオフ法で金属電極4を形成した(図5e).

【0015】密治性改善層9及びパリア層8のCrO。 郊膜で挟持された引出し電極5の金属電極4に対する接 触抵抗Rを測定した結果を表1に示す。この場合にも、 ペーク温度に拘わらず低い接触抵抗が維持された。ま た、基根/Cァ蒋暾及び基板/CァOェ蒋騏/Cァ蒋嶼 を基盤目密菪試験に供して密菪性を調査したところ、そ れぞれ75/100及び100/100の試験結果が得 られ、CrOx 蒋膑を介在させることによって基板1に 対する引出し登穫5の密着性が向上することが利った。

[0016]

【比較例】パリア層8及び密着性改善層9を形成しない ことを除き、実施例1と同じ方法で引出し電極5を透明

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成された引出し電極5の企風電極4に対する接触抵抗R を測定したところ、表1にみられるようにベークにより 接触抵抗が増加し、なかでも250℃を超えるペーク温 度では接触抵抗の増加が碧しかった。接触抵抗の窘しい* ⇒ 増加は、引出し遺極5の表面に変質層7が生成した結果 であり、駆励時に有機エレクトロルミネッセンス表示デ バイスを昇温させる原因にもなる。

[0017]

表1:Cr引出し電磁の層構成及びペーク温度が金属電極との接触抵抗(Q)に及ぼす影響

引出し電極の層構造	ペーク温度 (℃)					
	20	150	200	250	300	350
Cr/CrOx	2.8×10 ³	3.9×10 ³	8.6×10 ¹	2.5×10 ¹	1.7×10 ¹	8.0×10 ¹
CrOx/Cr/ CrOx	2.9×10 ³	3.3×10 ¹	8.7×10 ¹	2.5×10 ¹	1.7×10 ¹	7.9×10 ¹
パリア層なし	6.2×10 ¹	2.3×10 ²	1.1×10 ²	6.3×10 ²	4.8×10 ³	2.8×10 ⁵

[0018]

【発明の効果】以上に説明したように、本発明の有機工 レクトロルミネッセンス表示デバイスは、透明基板の上 に設けられる引出し単極の表層にパリア窟を形成してい るので、層間絶縁膜形成時等の際に加熱されても引出し 20 電極の表面に変質層が生じることがない。 したがって、 引出し電極と金属電極との接触抵抗が低く維持され、比 鮫的低い意圧で駆動することが可能な有機エレクトロル ミネッセンス表示デバイスとなる。また、密若性改善層 を介して引出し電極を形成するとき、透明基板に対する 引出し危極の密増性が向上し、引出し低極と金属低極と の良好な導流状態が維持される。

【図面の簡単な説明】

【図1】 層間絶録膜を設けた有機エレクトロルミネッダ

(図1)

【符号の説明】 1:透明基板

2:透明電極 3:有機発光層 6: 層間絶縁膜 4:金属電極 5:引出し電極 8:パリア層

7: 变黄層

※センス表示デバイスの層構造を示す断頭図

触抵抗が上昇することを説明する図

示デバイスを作製する工程図

ポデバイスを作製する工程図

【図2】 引出し電極の表面に生成した変質層により接

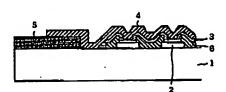
【図3】 本発明に従ってパリア暦を形成した有機エレ クトロルミネッセンス表示デバイスの層構造を示す断面

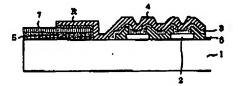
【図4】 実施例1の有機エレクトロルミネッセンス芸

【図5】 実施例2の有機エレクトロルミネッセンス表

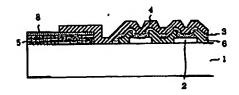
9:密若性改善層

[图2]



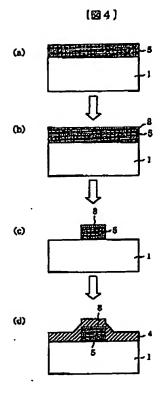


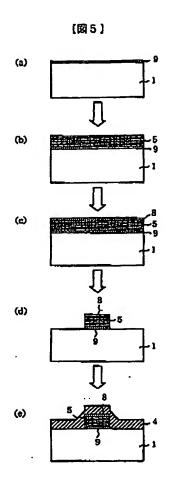
[図3]



(5)

特開2001-351778





[Detailed Description of the Invention] [0001]

[Industrial Application] This invention maintains the contact resistance to an electrode low, and relates to the organic electroluminescence display device which can be driven on a comparatively low electrical potential difference.

[0002]

[Description of the Prior Art] As a electrochromatic display which replaces the Braun tube, a liquid crystal display panel is various and has spread. However, since a liquid crystal display panel is a method which carries out image display with the light which passed the liquid crystal layer from the back light, an image may become hard to see with the include angle to see or surrounding intensity. An image is observed by visibility sufficient also in a dark place, without the display panel using this point and the electroluminescent element which displays a need image by field luminescence changing the conspicuousness of an image with an observation include angle.

[0003] Although various inorganic materials and organic materials are known as an electroluminescence ingredient, the organic material which carries out high brightness luminescence with low power attracts attention, the structure (drawing 1) to which the display device which used an organic luminescent material carried out the laminating of two or more stripe-like metal electrodes 4 which intersect perpendicularly with two or more stripe-like transparent electrodes 2, the organic luminous layer 3, and a transparent electrode 2 to the transparence substrate 1 one by one -- **** -- it is. On the transparence substrate 1, the laminating of the cash-drawer electrode 5 which flows in a metal electrode 4 is carried out.

[0004] If a drive current is supplied to the predetermined location on XY matrix formed with a transparent electrode 2 and a metal electrode 4, the electron from the hole side from an anode plate side and a cathode side will recombine by the organic luminous layer 3, an organic emitter molecule will be excited, and light will be emitted in the shape of a field. Luminescence is taken out outside through a transparent electrode 2 and the transparence substrate 1. The short circuit of the transparent electrode 2 and metal electrode 4 resulting from the level difference of the transparent electrode 2 edge section was prevented, and these people introduced by JP,8-315981,A that a clear image was displayed, when forming an interlayer insulation film 6 in this organic electroluminescence display device.

[0005]

[Problem(s) to be Solved by the Invention] An interlayer insulation film 6 applies for example, a silica distribution paste by the predetermined pattern, and is formed of BEKU. Or a polyimide system coating is applied and it is formed as a stable polymer coat by making a bridge construct by heat treatment. In any case, BEKU is needed for formation of an interlayer insulation film 6, and it is easy to deteriorate the front face of the cash-drawer electrode 5. Although the metal with resistance low as an ingredient of the cash-drawer electrode 5 is used, if the deterioration layer 7 (drawing 2) generates on a front face by BEKU, the contact resistance R with a metal electrode 4 will increase. Consequently, a high electrical potential difference is needed for the drive of an organic electroluminescence display device, and degradation of the organic electroluminescence display device by the temperature up is also promoted. [0006]

[Means for Solving the Problem] By being thought out that such a problem should be solved and forming a barrier layer in the front face of a cash-drawer electrode, this invention is pulled out at the time of heating of BEKU etc., prevents deterioration of an electrode, as a result the rise of the contact resistance of a cash-drawer electrode and a metal electrode, and aims at offering the organic electroluminescence display device which can be driven on a comparatively low electrical potential difference. It pulls out, a barrier layer is formed in the surface of an electrode, and the organic electroluminescence display device of this invention is characterized by said cash-drawer electrode touching the metal electrode which was prepared in the transparence substrate and by which the laminating was carried out on the transparence substrate through the organic luminous layer through said barrier layer in order to attain the purpose.

[0007] A cash-drawer electrode is formed with the nature electrical conducting material of a metal chosen from Cr, aluminum, Cu, Ag, Au, Pt, Pd, nickel, Mo, Ta, Ti, W, C, Fe, In, Ag-Mg, and Zn. A barrier layer is formed with the good refractory metal, the noble metals, the oxide, nitride, or acid nitride of heat-resistant deterioration nature. Moreover, a barrier layer can be formed also as a thin layer of the

oxide of the nature electrical conducting material for cash-drawer electrodes of a metal, a nitride, or an acid nitride. Furthermore, after forming the thin layer of the oxide of the nature electrical conducting material for cash-drawer electrodes of a metal, a nitride, or an acid nitride as an adhesion improvement layer, a cash-drawer electrode may be formed on a transparence substrate. The barrier layer formed in the surface of a cash-drawer electrode is processible using the same etching reagent with the cash-drawer electrode containing an adhesion improvement layer.

[Embodiment of the Invention] The organic electroluminescence display device according to this invention has covered the front face of the cash-drawer electrode 5 with the barrier layer 8, as shown in drawing 3. Others have the almost same layer structure as the conventional organic electroluminescence display device (drawing 1). By formation of the barrier layer 8, it pulls out at the time of interlayer insulation film 6 formation; the front face of an electrode 5 is protected thermally, and generation of the deterioration layer 7 is prevented. A metal or alloys, such as Cr, aluminum, Cu, Ag, Au, Pt, Pd, nickel, Mo, Ta, Ti, W, C, Fe, and In with low resistance, Ag-Mg, and Zn, are used for the ingredient of the cash-drawer electrode 5. These metals or alloys are formed on the transparence substrate 1 as a thin film by approaches, such as vacuum evaporationo, sputtering, and ion plating.

[0009] As an ingredient of the barrier layer 8, the refractory metal which does not deteriorate in the case of heat treatments, such as BEKU at the time of interlayer insulation film 6 formation, noble metals, the oxide, the nitride, the acid nitride, etc. are suitable. When forming the barrier layer 8 by the thickness (specifically 10-100A, preferably 10-500A, the optimal 10-200A) of extent which does not bar conductivity, the ingredient classified into the insulating material is also usable. Specifically, there are oxides, such as CrOx, AlOx, MoOx, FeOx, NiOx, AgOx, ZnOx, TaOx, WOx, SiOx, SnOx, CrNx, SiNx, CrNxOy, TiCx, TaSnOx, and TaSnOxNy, a nitride, an acid nitride, etc. other than metallic materials, such as Au, Pt, Pd, W, Moxetc recellent in deterioration-proof nature?

[0010] Moreover, it is also possible to use the metal used for the cash-drawer electrode 5 or the oxide of an alloy, a nitride, or an acid nitride for the barrier layer 8. In the culmination which forms the cashdrawer electrode 5 in this case, by introducing oxygen and/or nitrogen into an ambient atmosphere, the barrier layers 8, such as a metallic oxide, a metal nitride, and a metal acid nitride, pull out, and it is formed on an electrode 5. The barrier layer 8 formed with the oxide, nitride, or acid nitride of the same metal as the cash-drawer electrode 5 can carry out patterning using the same etching reagent as the cashdrawer electrode 5 in many cases. Although the barrier layer 8 formed in the surface of the cash-drawer electrode 5 is effective when preventing generation of the deterioration layer 7 at the time of the BEKU processing which forms an interlayer insulation film 6, also not only in the time of BEKU processing but in other heat treatment processes, generation of the deterioration layer 7 is controlled effectively, and the contact resistance of the cash-drawer electrode 5 to a metal electrode 4 is maintained to lower order. [0011] Furthermore, if sputtering of the Cr is carried out in advance of formation of the cash-drawer electrode 5 in Ar ambient atmosphere which introduced oxygen, the thin film of the quality of CrOx will be formed on the transparence substrate 1. The nature thin film of CrOx works as an adhesion improvement layer 9 (drawing 5) which raises the adhesion of the cash-drawer electrode 5 to a substrate 1. Patterning also of the nature thin film of CrOx can be carried out using the same etching reagent as the cash-drawer electrode 5. [0012]

[Example 1] After forming a transparent electrode 2 in the transparence substrate 1, the pure Cr thin film of 2900A of thickness was formed by sputtering which used Cr for the target in Ar ambient atmosphere (drawing 4 a). Subsequently, introduced oxygen into Ar ambient atmosphere, and pulled out CrOx, it was made to deposit on an electrode 5, and the CrOx thin film of 100A of thickness was formed (drawing 4 b). It etched into the transparence substrate 1 after the CrOx thin film was formed using the etching reagent which contains cerium-nitrate Amon after forming a resist pattern, and by exfoliating a resist, patterning of a pure Cr thin film and the CrOx thin film was carried out, and the cash-drawer electrode 5 and the barrier layer 8 of a predetermined configuration were formed (drawing 4 c). Since the same etching reagent draws out and patterning of the both sides of an electrode 5 and the barrier layer 8 of a predetermined-configuration can be created at a simple process.

[0013] Subsequently, after BEKU [inserting the transparence substrate 1 in oven and] in 150-300 degrees C, the metal electrode 4 was formed by the lift-off method which vapor-deposits aluminum (drawing 4 d). The result of having pulled out with the formed metal electrode 4 and having measured

the contact resistance R with an electrode 5 is shown in Table 1. The low contact resistance which formed CrOx as a barrier layer 8 and which pulled out, was maintaining low contact resistance to the metal electrode 4 in the electrode 5 also after BEKU, and was mostly fixed with the baking temperature of 200 degrees C or more was shown so that clearly from Table 1. [0014]

[Example 2] In order to raise the adhesion of the cash-drawer electrode 5 to the transparence substrate 1, the CrOx thin film (adhesion improvement layer 9) of 100A of thickness was formed on the transparence substrate 1 by sputtering which targeted Cr in Ar ambient atmosphere which introduced oxygen (drawing 5 a). subsequently, oxygen -- it replaced with free Ar ambient atmosphere, and sputtering of the Cr was carried out, Cr thin film (cash-drawer electrode 5) of 2800A of thickness was made to deposit (drawing 5 b), sputtering of the oxygen was again introduced and carried out to Ar ambient atmosphere, and the CrOx thin film (barrier layer 8) of 100A of thickness was formed on the metal electrode 4 (drawing 5 c). Patterning of each thin film was carried out by the same etching approach as an example 1, and the cash-drawer electrode 5 which equipped the lower layer with the adhesion improvement layer 9, and equipped the surface with the barrier layer 8 was formed (drawing 5 d). Subsequently, after BEKU [inserting the transparence substrate 1 in oven and] in 150-300 degrees C, the metal electrode 4 was formed by the lift-off method which vapor-deposits aluminum (drawing 5 e).

[0015] The result of having pulled out and having measured the contact resistance R to the metal electrode 4 of an electrode 5 pinched with the CrOx thin film of the adhesion improvement layer 9 and the barrier layer 8 is shown in Table 1. Also in this case, low contact resistance was maintained irrespective of baking temperature. Moreover, when the squares adherence test was presented with the substrate / Cr thin film, and the substrate / CrOx thin film / Cr thin film and adhesion was investigated, the test result of 75/100 and 100/100 was obtained, respectively, and by making a CrOx thin film intervene showed that the adhesion of the cash-drawer electrode 5 to a substrate 1 improved.

[Comparative Example(s)] After pulling out by the same approach as an example 1 and forming an electrode 5 on the transparence substrate 1 except for not forming the barrier layer 8 and the adhesion improvement layer 9, the metal electrode 4 was made to deposit. When it pulled out and the formed contact resistance R to the metal electrode 4 of an electrode 5 was measured, in the baking temperature which contact resistance increases by BEKU so that it may see in Table 1, and exceeds 250 degrees C especially, the increment in contact resistance was remarkable. The remarkable increment in contact resistance is the result of the deterioration layer 7 generating on the front face of the cash-drawer electrode 5, and also becomes the cause to which the temperature up of the organic electroluminescence display device is carried out at the time of a drive.

[0017]	 		
[0017] [X] ID=000003		·	
	<u>-</u> .		

[0018]

[Effect of the Invention] As explained above, since the organic electroluminescence display device of this invention forms the barrier layer in the surface of the cash-drawer electrode prepared on a transparence substrate, even if it is heated in the cases at the time of interlayer insulation film formation etc., it is pulled out, and a deterioration layer does not produce it on the surface of an electrode. Therefore, the contact resistance of a cash-drawer electrode and a metal electrode is maintained low, and it becomes the organic electroluminescence display device which can be driven on a comparatively low

electrical potential difference. Moreover, when pulling out through an adhesion improvement layer and
forming an electrode, the adhesion of the cash-drawer electrode to a transparence substrate improves,
and the good switch-on of a cash-drawer electrode and a metal electrode is maintained.

[Translation done.]

[Detailed Description of the Invention]

[0001]

[Industrial Application] This invention maintains the contact resistance to an electrode low, and relates to the organic electroluminescence display device which can be driven on a comparatively low electrical potential difference.

[0002]

[Description of the Prior Art] As a electrochromatic display which replaces the Braun tube, a liquid crystal display panel is various and has spread. However, since a liquid crystal display panel is a method which carries out image display with the light which passed the liquid crystal layer from the back light, an image may become hard to see with the include angle to see or surrounding intensity. An image is observed by visibility sufficient also in a dark place, without the display panel using this point and the electroluminescent element which displays a need image by field luminescence changing the conspicuousness of an image with an observation include angle.

[0003] Although various inorganic materials and organic materials are known as an electroluminescence ingredient, the organic material which carries out high brightness luminescence with low power attracts attention, the structure (drawing 1) to which the display device which used an organic luminescent material carried out the laminating of two or more stripe-like metal electrodes 4 which intersect perpendicularly with two or more stripe-like transparent electrodes 2, the organic luminous layer 3, and a transparent electrode 2 to the transparence substrate 1 one by one -- **** -- it is. On the transparence substrate 1, the laminating of the cash-drawer electrode 5 which flows in a metal electrode 4 is carried out.

[0004] If a drive current is supplied to the predetermined location on XY matrix formed with a transparent electrode 2 and a metal electrode 4, the electron from the hole side from an anode plate side and a cathode side will recombine by the organic luminous layer 3, an organic emitter molecule will be excited, and light will be emitted in the shape of a field. Luminescence is taken out outside through a transparent electrode 2 and the transparence substrate 1. The short circuit of the transparent electrode 2 and metal electrode 4 resulting from the level difference of the transparent electrode 2 edge section was prevented, and these people introduced by JP,8-315981,A that a clear image was displayed, when forming an interlayer insulation film 6 in this organic electroluminescence display device.

[Problem(s) to be Solved by the Invention] An interlayer insulation film 6 applies for example, a silica distribution paste by the predetermined pattern, and is formed of BEKU. Or a polyimide system coating is applied and it is formed as a stable polymer coat by making a bridge construct by heat treatment. In any case, BEKU is needed for formation of an interlayer insulation film 6, and it is easy to deteriorate the front face of the cash-drawer electrode 5. Although the metal with resistance low as an ingredient of the cash-drawer electrode 5 is used, if the deterioration layer 7 (drawing 2) generates on a front face by BEKU, the contact resistance R with a metal electrode 4 will increase. Consequently, a high electrical potential difference is needed for the drive of an organic electroluminescence display device, and degradation of the organic electroluminescence display device by the temperature up is also promoted. [0006]

[Means for Solving the Problem] By being thought out that such a problem should be solved and forming a barrier layer in the front face of a cash-drawer electrode, this invention is pulled out at the time of heating of BEKU etc., prevents deterioration of an electrode, as a result the rise of the contact resistance of a cash-drawer electrode and a metal electrode, and aims at offering the organic electroluminescence display device which can be driven on a comparatively low electrical potential difference. It pulls out, a barrier layer is formed in the surface of an electrode, and the organic electroluminescence display device of this invention is characterized by said cash-drawer electrode touching the metal electrode which was prepared in the transparence substrate and by which the laminating was carried out on the transparence substrate through the organic luminous layer through said barrier layer in order to attain the purpose.

[0007] A cash-drawer electrode is formed with the nature electrical conducting material of a metal chosen from Cr, aluminum, Cu, Ag, Au, Pt, Pd, nickel, Mo, Ta, Ti, W, C, Fe, In, Ag-Mg, and Zn. A barrier layer is formed with the good refractory metal, the noble metals, the oxide, nitride, or acid nitride of heat-resistant deterioration nature. Moreover, a barrier layer can be formed also as a thin layer of the

oxide of the nature electrical conducting material for cash-drawer electrodes of a metal, a nitride, or an acid nitride. Furthermore, after forming the thin layer of the oxide of the nature electrical conducting material for cash-drawer electrodes of a metal, a nitride, or an acid nitride as an adhesion improvement layer, a cash-drawer electrode may be formed on a transparence substrate. The barrier layer formed in the surface of a cash-drawer electrode is processible using the same etching reagent with the cash-drawer electrode containing an adhesion improvement layer.

[0008]

[Embodiment of the Invention] The organic electroluminescence display device according to this invention has covered the front face of the cash-drawer electrode 5 with the barrier layer 8, as shown in drawing 3. Others have the almost same layer structure as the conventional organic electroluminescence display device (drawing 1). By formation of the barrier layer 8, it pulls out at the time of interlayer insulation film 6 formation, the front face of an electrode 5 is protected thermally, and generation of the deterioration layer 7 is prevented. A metal or alloys, such as Cr, aluminum, Cu, Ag, Au, Pt, Pd, nickel, Mo, Ta, Ti, W, C, Fe and In with low resistance, Ag-Mg, and Zn, are used for the ingredient of the cash-drawer electrode 5. These metals or alloys are formed on the transparence substrate 1 as a thin film by approaches, such as vacuum evaporationo, sputtering, and ion plating.

[0009] As an ingredient of the barrier layer 8, the refractory metal which does not deteriorate in the case of heat treatments, such as BEKU at the time of interlayer insulation film 6 formation, noble metals, the oxide, the nitride, the acid nitride, etc. are suitable. When forming the barrier layer 8 by the thickness (specifically 10-100A, preferably 10-500A, the optimal 10-200A) of extent which does not bar conductivity, the ingredient classified into the insulating material is also usable. Specifically, there are oxides, such as CrOx, AlOx, MoOx, FeOx, NiOx, AgOx, ZnOx, TaOx, WOx, SiOx, SnOx, CrNx, SiNx, CrNxOy, TiCx, TaSnOx, and TaSnOxNy, a nitride, an acid nitride, etc. other than metallic materials, such as Au, Pt, Pd, W, Mo, etc. excellent in deterioration-proof nature.

[0010] Moreover, it is also possible to use the metal used for the cash-drawer electrode 5 or the oxide of an alloy, a nitride, or an acid nitride for the barrier layer 8. In the culmination which forms the cashdrawer electrode 5 in this case, by introducing oxygen and/or nitrogen into an ambient atmosphere, the barrier layers 8, such as a metallic oxide, a metal nitride, and a metal acid nitride, pull out, and it is formed on an electrode 5. The barrier layer 8 formed with the oxide, nitride, or acid nitride of the same metal as the cash-drawer electrode 5 can carry out patterning using the same etching reagent as the cashdrawer electrode 5 in many cases. Although the barrier layer 8 formed in the surface of the cash-drawer electrode 5 is effective when preventing generation of the deterioration layer 7 at the time of the BEKU processing which forms an interlayer insulation film 6, also not only in the time of BEKU processing but in other heat treatment processes, generation of the deterioration layer 7 is controlled effectively, and the contact resistance of the cash-drawer electrode 5 to a metal electrode 4 is maintained to lower order. [0011] Furthermore, if sputtering of the Cr is carried out in advance of formation of the cash-drawer electrode 5 in Ar ambient atmosphere which introduced oxygen, the thin film of the quality of CrOx will be formed on the transparence substrate 1. The nature thin film of CrOx works as an adhesion improvement layer 9 (drawing 5) which raises the adhesion of the cash-drawer electrode 5 to a substrate 1. Patterning also of the nature thin film of CrOx can be carried out using the same etching reagent as the cash-drawer electrode 5. [0012]

[Example 1] After forming a transparent electrode 2 in the transparence substrate 1, the pure Cr thin film of 2900A of thickness was formed by sputtering which used Cr for the target in Ar ambient atmosphere (drawing 4 a). Subsequently, introduced oxygen into Ar ambient atmosphere, and pulled out CrOx, it was made to deposit on an electrode 5, and the CrOx thin film of 100A of thickness was formed (drawing 4 b). It etched into the transparence substrate 1 after the CrOx thin film was formed using the etching reagent which contains cerium-nitrate Amon after forming a resist pattern, and by exfoliating a resist, patterning of a pure Cr thin film and the CrOx thin film was carried out, and the cash-drawer electrode 5 and the barrier layer 8 of a predetermined configuration were formed (drawing 4 c). Since the same etching reagent draws out and patterning of the both sides of an electrode 5 and the barrier layer 8 of a predetermined configuration can be created at a simple process.

[0013] Subsequently, after BEKU [inserting the transparence substrate 1 in oven and] in 150-300 degrees C, the metal electrode 4 was formed by the lift-off method which vapor-deposits aluminum (drawing 4 d). The result of having pulled out with the formed metal electrode 4 and having measured

the contact resistance R with an electrode 5 is shown in Table 1. The low contact resistance which formed CrOx as a barrier layer 8 and which pulled out, was maintaining low contact resistance to the metal electrode 4 in the electrode 5 also after BEKU, and was mostly fixed with the baking temperature of 200 degrees C or more was shown so that clearly from Table 1. [0014]

[Example 2] In order to raise the adhesion of the cash-drawer electrode 5 to the transparence substrate 1, the CrOx thin film (adhesion improvement layer 9) of 100A of thickness was formed on the transparence substrate 1 by sputtering which targeted Cr in Ar ambient atmosphere which introduced oxygen (drawing 5 a). subsequently, oxygen -- it replaced with free Ar ambient atmosphere, and sputtering of the Cr was carried out, Cr thin film (cash-drawer electrode 5) of 2800A of thickness was made to deposit (drawing 5 b), sputtering of the oxygen was again introduced and carried out to Ar ambient atmosphere, and the CrOx thin film (barrier layer 8) of 100A of thickness was formed on the metal electrode 4 (drawing 5 c). Patterning of each thin film was carried out by the same etching approach as an example 1, and the cash-drawer electrode 5 which equipped the lower layer with the adhesion improvement layer 9, and equipped the surface with the barrier layer 8 was formed (drawing 5 d). Subsequently, after BEKU [inserting the transparence substrate 1 in oven and] in 150-300 degrees C, the metal electrode 4 was formed by the lift-off method which vapor-deposits aluminum (drawing 5 e).

[0015] The result of having pulled out and having measured the contact resistance R to the metal electrode 4 of an electrode 5 pinched with the CrOx thin film of the adhesion improvement layer 9 and the barrier layer 8 is shown in Table 1. Also in this case, low contact resistance was maintained irrespective of baking temperature. Moreover, when the squares adherence test was presented with the substrate / Cr thin film, and the substrate / CrOx thin film / Cr thin film and adhesion was investigated, the test result of 75/100 and 100/100 was obtained, respectively, and by making a CrOx thin film intervene showed that the adhesion of the cash-drawer electrode 5 to a substrate 1 improved. [0016]

[Comparative Example(s)] After pulling out by the same approach as an example 1 and forming an electrode 5 on the transparence substrate 1 except for not forming the barrier layer 8 and the adhesion improvement layer 9, the metal electrode 4 was made to deposit. When it pulled out and the formed contact resistance R to the metal electrode 4 of an electrode 5 was measured, in the baking temperature which contact resistance increases by BEKU so that it may see in Table 1, and exceeds 250 degrees C especially, the increment in contact resistance was remarkable. The remarkable increment in contact resistance is the result of the deterioration layer 7 generating on the front face of the cash-drawer electrode 5, and also becomes the cause to which the temperature up of the organic electroluminescence display device is carried out at the time of a drive.

表1:Cr引出し電極の層構成及びベーク温度が金属電極との接触抵抗 (Ω) に及ぼす影響

引出し電極 の層構造	ベ ー ク 温 度 (℃)					
	20	150	200	250	300	350
Cr/CrOx	2.8×10 ³	3.9×10 ³	8.6×10 ¹	2.5×10 ¹	1.7×10 ¹	8.0×10 ¹
CrOx / Cr / CrOx	2.9×10 ³	3.3×10 ¹	8.7×10 ¹	2.5×10 ¹	1.7×10 ¹	7.9×10 ¹
バリア層なし	6.2×10 ¹	2.3×10 ²	1.1×10 ²	6.3×10 ²	4.8×10 ³	2.8×10 ⁵

[0018]

[Effect of the Invention] As explained above, since the organic electroluminescence display device of this invention forms the barrier layer in the surface of the cash-drawer electrode prepared on a transparence substrate, even if it is heated in the cases at the time of interlayer insulation film formation etc., it is pulled out, and a deterioration layer does not produce it on the surface of an electrode. Therefore, the contact resistance of a cash-drawer electrode and a metal electrode is maintained low, and it becomes the organic electroluminescence display device which can be driven on a comparatively low

electrical potential difference. Moreover, when pulling out through an adhesion improvement	ent layer and
forming an electrode, the adhesion of the cash-drawer electrode to a transparence substrate	improves,
and the good switch-on of a cash-drawer electrode and a metal electrode is maintained.	
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[Translation done.]

DESCRIPTION OF DRAWINGS

[Brief Description of the Drawings]

<u>[Drawing 1]</u> The sectional view showing the layer structure of the organic electroluminescence display device which prepared the interlayer insulation film

[Drawing 2] Drawing explaining contact resistance going up by the deterioration layer generated on the front face of a cash-drawer electrode

[Drawing 3] The sectional view showing the layer structure of the organic electroluminescence display device which formed the barrier layer according to this invention

[Drawing 4] Process drawing which produces the organic electroluminescence display device of an example 1

[Drawing 5] Process drawing which produces the organic electroluminescence display device of an example 2

[Description of Notations]

1: Transparence substrate 2: Transparent electrode 3: Organic luminous layer 4: Metal electrode 5: Cash-drawer electrode 6: Interlayer insulation film 7: Deterioration layer 8: Barrier layer 9: Adhesion improvement layer